**Problem 5.1** To study neutron absorption cross sections at very low energies, one must often slow down (moderate) energetic ( $\approx 1\,\mathrm{MeV}$ ) neutrons that are produced in reactors. Show that paraffin would be a better moderator than aluminum, by specifically calculating the maximum energy that a  $1\,\mathrm{MeV}$  neutron can transfer in a collision with a proton (within paraffin) as opposed to that with an Al nucleus.

As we saw in Problem 2.9 (see Eq. (2.48)), in a head-on collision with a target nucleus of mass number A, a neutron scatters backwards  $(\theta = \pi)$  with an energy

$$E_n = \left(\frac{A-1}{A+1}\right)^2 E_0,\tag{5.1}$$

where  $E_0$  represents the energy of the incident neutron. If the target is paraffin (essentially a chain of  $CH_2$  units), then ignoring the presence of C, for A=1 we get

$$E_n^{\text{(paraffin)}} = \left(\frac{A-1}{A+1}\right)^2 E_0 = \left(\frac{1-1}{1+1}\right)^2 E_0 = 0.$$
 (5.2)

On the other hand, if the target nucleus is aluminum ( $^{27}Al^{13}$ ), we have A=27, leading to

$$E_n^{\text{(aluminum)}} = \left(\frac{A-1}{A+1}\right)^2 E_0 = \left(\frac{27-1}{27+1}\right)^2 E_0$$
  
  $\approx 0.86E_0.$  (5.3)

Thus, we see that paraffin is a much better moderator, since after one collision a neutron can lose all its energy.