#### CHAPTER 1 Rutherford Scattering

Lecture Notes For PHYS 415 Introduction to Nuclear and Particle Physics

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# **Rutherford Scattering**

#### Scattered $\boldsymbol{\alpha}$ particles from thin foils



# Effect of atomic electrons

Mass of  $\alpha \approx 4 \times 10^3 \text{ MeV/c}^2$ 

Mass of  $e^- \approx 0.5 \text{ MeV/c}^2$ 

Electrons cannot cause large deviations.

- Further, if mass of the atom is spread throughout the atomic volume, only minor deflections would be observed.
- Let's analyze more carefully ...



**Momentum conservation:**  $m_{\alpha}\vec{v}_{0} = m_{\alpha}\vec{v}_{\alpha} + m_{t}\vec{v}_{t}$ 

 $\Rightarrow \vec{v}_0 = \vec{v}_\alpha + \frac{m_t}{m_\alpha} \vec{v}_t$ 

Energy conservation:

$$\frac{1}{2}m_{\alpha}v_{0}^{2} = \frac{1}{2}m_{\alpha}v_{\alpha}^{2} + \frac{1}{2}m_{t}v_{t}^{2}$$
$$\Rightarrow v_{0}^{2} = v_{\alpha}^{2} + \frac{m_{t}}{m_{\alpha}}v_{t}^{2}$$

# Analysis, cont'd.

Combining these equations, gives

$$v_t^{2} \left( 1 - \frac{m_t}{m_\alpha} \right) = 2 \vec{v}_\alpha \cdot \vec{v}_t$$

- For  $m_{\alpha} >> m_{t}$ , LHS > 0 and motion of  $\alpha$  is along incident direction.
- For  $m_{\alpha} << m_{\rm t}$ , LHS < 0 and motion of  $\alpha$  is along backward direction.
- For electrons as target, the first condition applies and hence backward scattering does not occur.

# Other comments on the scattering

- If mass and charge are evenly distributed, the theory predicts only small scattering angles with prob(θ >90) ~ 10<sup>-3000</sup>
- Experiment revealed  $prob(\theta > 90^{\circ}) \sim 10^{-4}$
- If one assumes scattering from a single electron in the material,  $\theta_{max}$ =0.016°. Considering the # of atoms (~2300) across the thickness of the foil, this number increases to ( $\sqrt{2300}$ ) × 0.016° = 0.8° (assuming one electron per atom can scatter).
- Even if the  $\alpha$  particle scattered from all 79 electrons in each atom of gold,  $\langle \theta \rangle_{total} = 6.8^{\circ}$
- Conclusion: Only a massive concentrated "nuclear" center can give rise to observed large angle scattering.

# **Coulomb Scattering**

- Energy/Momentum conservation gives the correct asymptotic values for the particles involved in the scattering.
- To go further, we must consider the Coulomb force between the  $\alpha$  and the atomic nucleus.
- Both are positively charged, so the force is repulsive.
- First, an aside on **UNITS** ...

#### Units Conventions (with some exceptions)

Mass, Energy, Momentum:

- □ eV/c², eV, eV/c
- □ MeV/c<sup>2</sup>, MeV, MeV/c
- Otherwise: "cgs" units: cm, g, s
- Electromagnetic formulae take on simpler forms, but the units of charge are now:

 $1esu = 1Stat-Coulomb = 3.34 \times 10^{-10} C$ 

 $\Rightarrow$  e = 4.8×10<sup>-10</sup> esu

# **Coulomb Scattering Analysis**



- Assume nucleus is infinitely massive  $\Rightarrow$  no recoil
- Use non-relativistic kinematics
- Target is thin  $\Rightarrow$  only one scattering
- Projectile and nucleus are point-like objects
- Consider only the Coulomb force

#### Scattering from a central potential V(r)

$$\frac{d\chi}{dt} = \frac{\ell}{mr^2}.$$
(1.16)

The energy is identical at every point of the trajectory, and can be written as

$$E = \frac{1}{2} m \left(\frac{dr}{dt}\right)^2 + \frac{1}{2} mr^2 \left(\frac{d\chi}{dt}\right)^2 + V(r)$$
$$= \frac{1}{2} m \left(\frac{dr}{dt}\right)^2 + \frac{1}{2} mr^2 \left(\frac{\ell}{mr^2}\right)^2 + V(r),$$

or 
$$\frac{1}{2}m\left(\frac{dr}{dt}\right)^2 = E - \frac{\ell^2}{2mr^2} - V(r),$$
  
or  $\frac{dr}{dt} = -\left[\frac{2}{m}\left(E - V(r) - \frac{\ell^2}{2mr^2}\right)\right]^{\frac{1}{2}}.$  (1.17)

$$\frac{dr}{dt} = -\left[\frac{2}{m}\frac{\ell^2}{2mr^2}\left\{\frac{2mEr^2}{\ell^2}\left(1-\frac{V(r)}{E}\right)-1\right\}\right]^{\frac{1}{2}} \\
= -\frac{\ell}{mr}\left[\frac{r^2}{b^2}\left(1-\frac{V(r)}{E}\right)-1\right]^{\frac{1}{2}} \\
= -\frac{\ell}{mrb}\left[r^2\left(1-\frac{V(r)}{E}\right)-b^2\right]^{\frac{1}{2}}.$$
(1.18)

$$\begin{split} d\chi &= \frac{\ell}{mr^2} \, dt = \frac{\ell}{mr^2} \, \frac{dt}{dr} \, dr \\ &= -\frac{\ell}{mr^2} \frac{dr}{\frac{\ell}{mrb} \left[ r^2 \left( 1 - \frac{V(r)}{E} \right) \, - b^2 \right]^{\frac{1}{2}}}, \end{split}$$

or 
$$d\chi = -\frac{bdr}{r\left[r^2\left(1-\frac{V(r)}{E}\right)-b^2\right]^{\frac{1}{2}}}$$
. (1.19)

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$$\int_{0}^{\chi_{0}} d\chi = -\int_{\infty}^{r_{0}} \frac{bdr}{r \left[r^{2} \left(1 - \frac{V(r)}{E}\right) - b^{2}\right]^{\frac{1}{2}}},$$
  
or  $\chi_{0} = b \int_{r_{0}}^{\infty} \frac{dr}{r \left[r^{2} \left(1 - \frac{V(r)}{E}\right) - b^{2}\right]^{\frac{1}{2}}}.$  (1.20)

The point of closest approach is determined by noting that, as the particle approaches from infinity, its velocity decreases continuously (assuming the repulsive potential for the case of an  $\alpha$ -particle approaching a nucleus), until the point of closest approach, where the radial velocity  $\left(\frac{dr}{dt}\right)$  vanishes and subsequently changes sign. That is, beyond this point, the velocity of the particle increases again. Therefore, at the distance of closest approach, when  $r = r_0$ , both the radial and the absolute velocities attain a minimum, and we have

$$\left. \frac{dr}{dt} \right|_{r:=r_0} = 0,$$

which, from Eqs. (1.17) and (1.18), means that

$$E - V(r_0) - \frac{\ell^2}{2mr_0^2} = 0,$$
  
or  $r_0^2 \left(1 - \frac{V(r_0)}{E}\right) - b^2 = 0.$  (1.21)

# Scattering angle as a function of the impact parameter

Thus, given a specific form of the potential, we can determine  $r_0$ , and therefore  $\chi_0$ , as a function of the impact parameter b.<sup>3</sup> Defining the scattering angle  $\theta$  as the change in the asymptotic angles of the trajectory, we get

$$\theta = \pi - 2\chi_0 = \pi - 2b \int_{r_0}^{\infty} \frac{dr}{r \left[r^2 \left(1 - \frac{V(r)}{E}\right) - b^2\right]^{\frac{1}{2}}}.$$
 (1.22)

#### Impact parameter

Coulomb force:

 $V(r) = \frac{ZZ'e^2}{r}$ 

This leads to a relation between the impact parameter, *b*, and the incident energy,  $E=1/2 \text{ mv}_0^2$ , and scattering angle,  $\theta$ :

$$b = \frac{ZZ'e^2}{2E}\cot\frac{\theta}{2}$$

Small *b* corresponds to large angle scattering  $\Rightarrow$  large  $\theta$  is possible since nucleus is treated as point object and we get a large force close to the nucleus.

# Distance of closest approach

• It can also be shown that the distance of closest approach,  $r_0$ , is given by

$$r_{0} = \frac{ZZ'e^{2}}{2E} \left( 1 + \sqrt{1 + \frac{4b^{2}E^{2}}{\left(ZZ'e^{2}\right)^{2}}} \right) \xrightarrow{E \to \infty} b$$

- The previous slide shows that for non-zero  $\theta$ ,  $b \rightarrow 0$  as  $E \rightarrow \infty$ . Therefore  $r_0 \rightarrow 0$  as  $E \rightarrow \infty$ .
- Thus, at high enough energy we can approach the nucleus as closely as we wish. The assumption of a point-like nucleus can then be tested.

# **Scattering Cross Section**

In an actual experiment, detectors are positioned to cover a range of scattering angles and, therefore, impact parameters.



#### **Differential Cross Section**

Effective area for scattering into  $d\theta$ :

$$\Delta \sigma(\theta, \phi) = b \ db \ d\phi = -\frac{d\sigma}{d\Omega}(\theta, \phi) d\Omega = -\frac{d\sigma}{d\Omega}(\theta, \phi) \sin\theta \ d\theta \ d\phi$$
$$\xrightarrow{\text{no } \phi \text{ dependence}} \frac{d\sigma}{d\Omega}(\theta) = -\frac{b}{\sin\theta} \ \frac{db}{d\theta}$$

Entire effect of scattering is given in the  $\theta$  dependence of the differential cross section, or of the yield.

We can get the explicit dependence on observable parameters for the case of Rutherford scattering (i.e. Coulomb force) ...

#### **Rutherford Cross Section**



# **Total Cross Section**

We can get the total cross section by integrating:

$$\sigma_{\text{TOT}} = \int \frac{d\sigma}{d\Omega}(\theta) d\Omega = 2\pi \int_0^\pi d\theta \sin\theta \frac{d\sigma}{d\Omega}(\theta)$$
$$= 8\pi \left(\frac{ZZ'e^2}{4E}\right)^2 \int_0^1 d\left(\sin\frac{\theta}{2}\right) \frac{1}{\sin^3\frac{\theta}{2}} \to \infty$$

The infinite result reflects the infinite range of the Coulomb potential. Normally, the detectors exclude very small scattering angles.

# **Cross Section Units**

1 bn = 1 barn =  $10^{-24}$  cm<sup>2</sup> (Typical nuclear radius ~  $10^{-12}$  cm = 10 fm)

# Differential cross section, for example, may be expressed in **mb/sr** units

 $(4\pi \text{ sr} = \text{full solid angle about a point})$ 

 $(1 \text{ mb} = 1 \text{ millibarn} = 10^{-3} \text{ bn})$ 

#### **Measuring Cross Sections**

Consider *dn* particles per unit time, scattering into a solid angle  $d\Omega$  at a given  $\theta$  and  $\phi$ 



# Measuring Cross Sections, cont'd.

# nuclei (scatterers) per unit area:

$$\frac{N}{S} = \frac{\rho t}{A} A_0$$

$$\rho = \text{density of foil (g/cm^3)}$$

$$t = \text{thickness of foil (cm)}$$
where
$$A = \text{mass } \# (g/\text{mole})$$

$$A_0 = \text{Avogadro's } \# = 6.02 \times 10^{23} \text{ mole}^{-1}$$

$$S = \text{area of foil (cm^2)}$$

# incident particles per unit time =  $N_0$ 

# Measuring Cross Sections, cont'd.

The cross section has a statistical interpretation, though it's measured in units of area:

$$dn = N_0 \frac{N}{S} \frac{d\sigma}{d\Omega}(\theta, \phi) d\Omega = N_0 \frac{\rho t}{A} A_0 \frac{d\sigma}{d\Omega}(\theta, \phi) d\Omega$$

- Geiger and Marsden made such measurements and verified the Rutherford prediction.
- This gave convincing evidence of the hypothesis of a nuclear center, but provided no information about the nature of the nuclear force: the scattering is entirely due to the Coulomb repulsion, as the α particles never penetrated the nucleus.

# Laboratory vs. CoM Frame

- We have ignored nuclear recoil, supposing that the mass of the target was infinite.
- We can treat this 2-body scattering problem in terms of relative and center-of-mass coordinates.
- The scattering can then be separated for central potentials:
  - The CoM moves at constant velocity.
  - The relative motion can be treated exactly as before, except that the projectile mass is replaced by the "reduced mass".
- This is especially useful in treating colliding beam experiments.

# Definitions

Define 
$$\begin{cases} \vec{r} = \vec{r_1} - \vec{r_2} , \quad \vec{R}_{CM} = \frac{m_1 \vec{r_1} + m_2 \vec{r_2}}{m_1 + m_2} \\ M = m_1 + m_2 = \text{ total mass} \\ \mu = \frac{m_1 m_2}{m_1 + m_2} = \text{ reduced mass} \end{cases}$$



#### **Transforming between Frames**



# **Transformation Equations**

$$\begin{aligned} v_{CM} &= \dot{R}_{CM} = \frac{m_1 v_1}{m_1 + m_2} \\ \tilde{v}_1 &= v_1 - v_{CM} = \frac{m_2 v_1}{m_1 + m_2} \\ \tilde{v}_2 &= v_{CM} = \frac{m_1 v_1}{m_1 + m_2} \\ \tan \theta_{Lab} &= \frac{\sin \theta_{CM}}{\cos \theta_{CM} + \zeta} \quad \text{with } \zeta = \frac{m_1}{m_2} \\ \frac{d\sigma}{d\Omega_{Lab}} (\theta_{Lab}) &= \frac{d\sigma}{d\Omega_{CM}} (\theta_{CM}) \frac{d(\cos \theta_{CM})}{d(\cos \theta_{Lab})} \\ &= \frac{d\sigma}{d\Omega_{CM}} (\theta_{CM}) \frac{(1 + 2\zeta \cos \theta_{CM} + \zeta^2)^{3/2}}{|1 + \zeta \cos \theta_{CM}|} \end{aligned}$$

#### **Relativistic Variables**

- So far we have ignored relativity. For most modern nuclear and particle physics experiments, we must treat the kinematics relativistically.
- First, we need the center-of-mass velocity:

$$\frac{\vec{v}_{CM}}{c} = \vec{\beta}_{CM} = \frac{\vec{P}_1 + \vec{P}_2}{E_1 + E_2} c \xrightarrow{\text{in Lab frame}} \frac{\vec{P}_1 c}{E_1 + m_2 c^2}$$

Which gives:

$$\gamma_{CM} = \left(1 - \beta_{CM}^{2}\right)^{-1/2} = \frac{E_1 + m_2 c^2}{\left(m_1^2 c^4 + m_2^2 c^4 + 2E_1 m_2 c^2\right)^{1/2}}$$

#### **Relativistic Invariants**

Certain quantities are frame-independent; they can therefore be evaluated in any (convenient) frame. Examples are the "Mandelstam" variables, *s*, *t* and *u*.



$$s = (p_1 + p_2)^2 = (p_3 + p_4)^2$$
  

$$t = (p_1 - p_3)^2 = (p_2 - p_4)^2$$
  

$$u = (p_1 - p_4)^2 = (p_2 - p_3)^2$$
  
where the *p*'s are 4 - vectors  
and the squares imply 4 - vector dot products

# Total CoM Energy

For 2-body scattering, the total CoM energy is:

$$s = (E_1 + E_2)^2 - (\vec{P}_1 + \vec{P}_2)^2 c^2$$
  
=  $m_1^2 c^4 + m_2^2 c^4 + 2E_1 m_2 c^2$  (in Lab)  
=  $(E_{1CM} + E_{2CM})^2 = (E_{CM}^{TOT})^2$  (in CoM)

• Also: 
$$\gamma_{CM} = \frac{E_1 + m_2 c^2}{E_{CM}^{TOT}} = \frac{E_{Lab}^{TOT}}{E_{CM}^{TOT}}$$

 The total CoM energy may be regarded as the available energy to produce particles, since the motion of the CoM itself is unchanged after scattering.

#### **Four-Momentum Transfer**

The Mandelstam variable, t, is just the square of the fourmomentum transferred to the target:

$$t = \left(E_1^{\ f} - E_1^{\ i}\right)^2 - \left(\vec{P}_1^{\ f} - \vec{P}_1^{\ i}\right)^2 c^2$$
$$= \left(E_2^{\ f} - E_2^{\ i}\right)^2 - \left(\vec{P}_2^{\ f} - \vec{P}_2^{\ i}\right)^2 c^2$$

In elastic scattering viewed from the CoM, each particle's speed, and therefore energy, does not change. Therefore:

$$t = -\left(P_{1CM}^{f^{2}} + P_{1CM}^{i^{2}} - 2\vec{P}_{1CM}^{f} \cdot \vec{P}_{1CM}^{i}\right)c^{2}$$
$$= -2P_{CM}^{2}c^{2}\left(1 - \cos\theta_{CM}\right) < 0$$

# Feynman Diagrams

- From the definition of *t*, we can consider the scattering to take place by the exchange of a particle, of mass *m*, where *t* = *m*<sup>2</sup>.
- Since t < 0, the particle has an imaginary rest mass and is therefore called virtual.
- Feynman invented pictorial representations of scattering processes, in which each picture has a precise mathematical meaning: a Feynman diagram.

# Feynman Diagrams, cont'd.



- Such diagrams can be used to calculate scattering amplitudes and cross sections.
- Feynman invented them to calculate processes in QED (quantum electrodynamics) via the technique of perturbation theory.

#### Interpretation of Four-momentum Transfer

• Define 
$$q^2c^2 = -t$$
. We can show that:

$$q^2 = 2m_2 T_{2Lab}^f \xrightarrow{v_2 << c} (m_2 v_2)^2$$

Therefore q is related to the momentum transferred to the target and reflects the "hardness" of the collision. Long-range collisions (soft collisions) are characterized by small q and vice versa:

$$R \approx \frac{\hbar}{q}$$

■ Also, for small  $\theta_{CM}$ ,  $q^2 \approx P_{CM}^2 \theta_{CM}^2 \approx p_T^2$  = square of transverse momentum due to collision.

# **Back to Rutherford Scattering**

For a massive target, we can rewrite the Rutherford cross section as:

$$\frac{d\sigma}{dq^2} = \frac{4\pi \left(ZZ'e^2\right)^2}{v^2} \frac{1}{q^4}$$

- The q<sup>-4</sup> divergence reflects the 1/r dependence of the Coulomb potential.
- The average momentum transfer for all angles is small.

# Quantum Treatment of Rutherford Scattering

- So far the treatment of Rutherford scattering has been classical.
- We can calculate this process in QM using *Fermi's Golden Rule*. The transition probability is:

$$P = \frac{2\pi}{\hbar} |H_{fi}|^2 \rho(E_f)$$
  
where  $H_{fi} = \langle f | H | i \rangle = \int d^3 r \psi_f^*(\vec{r}) H(r) \psi_i(\vec{r})$ 

# Quantum Treatment, cont'd.

 Using plane waves for the incident and scattered particle and using the Coulomb potential energy for *H*, and defining wave vectors:

$$H_{fi} \approx \int d^3 r \exp\left(i\vec{k}'\cdot\vec{r}\right) V(r) \exp\left(-i\vec{k}\cdot\vec{r}\right) = \int d^3 r V(r) \exp\left(\frac{i}{\hbar}\vec{q}\cdot\vec{r}\right)$$

- This is the Fourier transform of V(r).
- After performing the integral and calculating the density of final states, we obtain the same expression as given before.
- The classical calculation gives the correct quantum mechanical result (when spin is ignored).