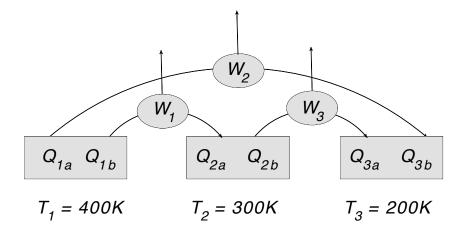
## Problem 5-7 solution



## Solution #1 (a)

We assume that the engines are Carnot engines, then

$$\begin{aligned} Q_{2a} &= \frac{T_2}{T_1}Q_{1b} \,=\, \frac{3Q_{1b}}{4}, \qquad Q_{3b} \,=\, \frac{T_3}{T_1}Q_{1a} \,=\, \frac{Q_{1a}}{2}, \qquad Q_{3a} \,=\, \frac{T_3}{T_2}Q_{2b} \,=\, \frac{2Q_{2b}}{3} \\ &\Rightarrow W_1 \,=\, Q_{1b} - Q_{2a} \,=\, \frac{Q_{1b}}{4}, \quad W_2 \,=\, Q_{1a} - Q_{3b} \,=\, \frac{Q_{1a}}{2}, \quad W_3 \,=\, Q_{2b} - Q_{3a} \,=\, \frac{Q_{2b}}{3} \end{aligned}$$

Also,

$$\begin{array}{lll} Q &=& Q_{1a} + Q_{1b} &=& 1200 J \\ W &=& W_1 + W_2 + W_3 &=& \frac{Q_{1a}}{2} \; + \; \frac{Q_{1b}}{4} \; + \frac{Q_{2b}}{3} \; = \; 200 J \end{array}$$

We get

$$W = \frac{Q_{1a}}{2} + \frac{Q_{1b}}{4} + \frac{Q_{2b}}{3} = \frac{Q_{1a} + Q_{1b}}{2} - \frac{Q_{1b}}{4} + \frac{Q_{2b}}{3} = \frac{Q}{2} - \frac{Q_{2a}}{3} + \frac{Q_{2b}}{3}$$

$$\Rightarrow Q_{2a} - Q_{2b} = 3(\frac{Q}{2} - W) = 1200J$$

 $\Rightarrow$  1200J is rejected at the second reservoir.

Next, from the overall conservation of energy

$$W = Q_{1a} + Q_{1b} - Q_{2a} + Q_{2b} - Q_{3a} - Q_{3b} = -Q_{3a} - Q_{3b} = 200J \Rightarrow (Q_{3a} + Q_{3b}) = -200J$$

 $\Rightarrow$  200J of heat is absorbed at the third reservoir

(b), (c)

$$\Delta S_1 \ = \ \frac{Q_{1a} + Q_{1b}}{T_1} \ = \ 3\frac{J}{K}, \quad \Delta S_2 \ = \ \frac{Q_{2a} - Q_{2b}}{T_2} \ = \ -4\frac{J}{K}, \quad \Delta S_3 \ = \ \frac{Q_{3a} + Q_{3b}}{T_3} \ = \ 1\frac{J}{K}$$

 $\Rightarrow \Delta S = \Delta S_1 + \Delta S_2 + \Delta S_3 = 0$  as it should be for a reversible process.

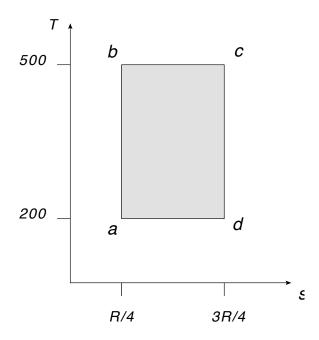
## Solution #2

Let us denote heats absorbed by the engine as  $Q_1, Q_2$ , and  $Q_3$  (so  $Q_1 = Q_{1a} + Q_{1b}$ ,  $Q_2 = -Q_{2a} + Q_{2b}$ ,  $Q_3 = -Q_{3a} - Q_{3b}$ ). If the process is reversible, change of entropy is zero so we get

$$\begin{aligned} \frac{Q_1}{T_1} + \frac{Q_2}{T_2} + \frac{Q_3}{T_3} &= 0 \quad \text{and} \quad Q_1 + Q_2 + Q_3 = W \\ \frac{Q_1}{T_1} + \frac{Q_2}{T_2} - \frac{Q_1 + Q_2}{T_3} + \frac{W}{T_3} &= 0 \quad \Rightarrow \quad 3 + \frac{Q_2}{300} - 6 + 1 &= 0 \quad \Rightarrow \quad Q_2 = -1200, \ Q_3 = 200 \end{aligned}$$

so 1200J is rejected at reservoir 2 and 200J is absorbed at reservoir 3.

## Problem 5-13 solution



(a): heat engine

(b)

$$Q_{a\to b} = 0,$$
  
 $Q_{b\to c} = T_2(s_2 - s_1) = 250R \simeq 2.1 \times 10^6 J,$   
 $Q_{c\to d} = 0,$   
 $Q_{d\to a} = T_1(s_1 - s_2) = -100R \simeq -8.3 \times 10^5 J.$ 

(c)

$$\eta = \frac{W}{Q_{b \to c}} = \frac{Q_{b \to c} + Q_{d \to a}}{Q_{b \to c}} = \frac{T_2 - T_1}{T_2} = 60\%$$

(d) In reverse

$$c = \frac{Q_{a \to d}}{Q_{b \to c} - Q_{a \to d}} = \frac{T_2}{T_2 - T_1} = \frac{2}{3} \simeq 67\%$$